

# Heisenberg's Uncertainty Principle

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- The very act of observing affects the particle being observed.
- Both momentum and position can't be known with infinite precision

Consider : particle in a 1-D box.

Desire : measure momentum component in the  $x$ -direction for a set of identical systems for which the particle is known to be in the lowest energy state.

i.e.  $n=1$   $\psi = A \sin \frac{n\pi x}{L}$

$$\hat{P}_x = -i\hbar \frac{d}{dx}$$

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Consider:  $\hat{p}_x \psi$

$$\hat{p}_x \psi = -i\hbar \frac{d}{dx} \left( A \sin \frac{\pi x}{L} \right)$$

$$= -i\hbar A \frac{\pi}{L} \cos \frac{\pi x}{L}$$

Operating on  $\sin \Rightarrow \cos$

$\therefore \psi$  not an eigenfunction  
of  $\hat{p}$

- measurements will not  
yield a precise result.

Need to use the mean value theorem  
to obtain the expectation value.

$$\langle P_x \rangle = \int_0^L \psi_1 \hat{P}_x \psi_1 dx$$

$$-i\hbar A \frac{\pi}{L} \cos \frac{\pi x}{L}$$

$$A = \left(\frac{2}{L}\right)^{1/2}$$

$$\langle P_x \rangle = \left(\frac{2}{L}\right)^{1/2} \left(\frac{2}{L}\right)^{1/2} \left(-i\hbar \frac{\pi}{L}\right) \int_0^L \sin \frac{\pi x}{L} \cos \frac{\pi x}{L} dx$$

$$= 0$$

propagation in both directions



But 1-D Particle in a Box

$$E_n = \frac{n^2 h^2}{8mL^2} \Rightarrow \frac{h^2}{8mL^2} \quad (n=1)$$

$$P_x^2 = \frac{h^2}{4L^2}$$

Hence  $P_x^2 = 2mE_1$  ( $\leftarrow n=1$ )

$$P_x^{(1)} = \pm \sqrt{2mE_1}$$

Ave. value  $P_x^{(1)} = 0$

Always some +, some -

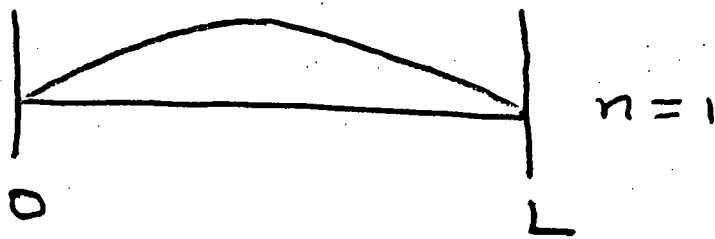
Never know which you will measure

Momentum An Uncertainty exists in ones Knowledge =  $2(2mE_1)^{1/2}$

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Now, Consider position

$\Psi_n \Rightarrow \Psi_1 =$  state of the particle



Uncertainty in position - particle  
is some where in the box

$$\Delta P_x \Delta X \approx 2(2mE_1)^{1/2} \cdot L$$

Product of the  
uncertainties

$$\geq 2\left(2m \frac{h^2}{8mL^2}\right)^{1/2} \cdot L$$

$$\geq h$$

More Exact  
Reasoning

$$\Delta P \Delta x \geq h/2$$